

Kinetic Resolution of Aliphatic 1,2-Diols by a Lipase-Catalyzed Sequential Acetylation¹

Fritz Theil*, Judith Weidner, Sibylle Ballschuh, Annamarie Kunath, and Hans Schick

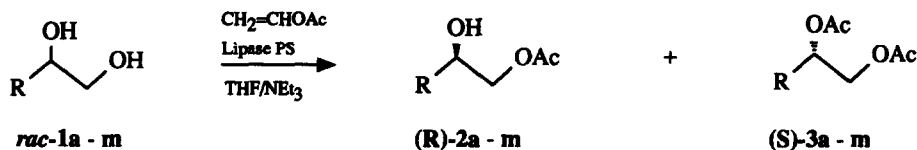
Centre of Selective Organic Synthesis, Rudower Chaussee 5, D(O)-1199 Berlin-Adlershof, Germany

Abstract: The kinetic resolution of 13 racemic aliphatic 1,2-diols (*rac*-1a-m) by means of a lipase-catalyzed sequential acetylation was investigated. The enantioselectivity of the 3-aryloxy-propane-1,2-diols *rac*-1a-k depends on the substitution pattern at the aryl ring.

Homochiral 1,2-diols are of increasing interest as synthetic intermediates for different classes of substances.² Thus, these compounds play an important role as target molecules in asymmetric synthesis with enzymes as chiral catalysts. It is known that lipases catalyze the regioselective acylation of aliphatic 1,2-diols at the primary hydroxy group³ with low enantioselectivity.⁴ To overcome this problem it is necessary to protect the primary hydroxy group with a bulky group.⁵

3-Aryloxy-propane-1,2-diols are building blocks for β -blockers.⁶ Derived⁷ and related⁸ compounds are used as substrates in bioconversions. The increasing number of papers in this field prompted us to publish our recent results in a preliminary manner.

It was our aim to use diols, which do not require a manipulation at the primary hydroxy group, as substrates in a lipase-catalyzed sequential acetylation.⁹ We could demonstrate the amplification of the enantioselectivity by application of this concept in the case of racemic *Mephesisin*^R (*rac*-1b) as substrate.^{4b} This reaction showed a moderate enantioselectivity ($E = 27$).¹⁰ In order to obtain information about the influence of the side chain of these diols on the enantioselectivity of the lipase-catalyzed sequential acetylation, the 2-methylphenoxy residue was replaced by other substituted aryloxy groups and by one alkyl, as well as one aryl substituent in the 2-position of the diol (Scheme 1).



Scheme 1

These lipase-catalyzed transesterifications were carried out as previously described.^{4b} The results of the kinetic resolutions are summarized in Table 1. In the presence of lipase *Amano PS*, the (S)-enantiomers of 1a-k are converted at a lower rate into the primary (R)-monoacetates 2a-k. The corresponding (R)-enantiomers of 1a-k are converted at a higher rate into the (S)-diacetates 3a-k. In general, the derivatives with substituents in the 4-position of the aromatic ring show significantly higher enantioselectivities than those with substituents in the 2-position. Such a clear relationship could not be observed for the compounds substituted in the 3-position of the aromatic ring. Furthermore, a substituent in the 4-position seems to be a

prerequisite for a high enantioselectivity of the resolution procedure (Table 1). The observed selectivities correspond with Kazlauskas' rule.¹¹

Acknowledgements: For a generous gift of lipase PS we would like to thank the Amano Pharmaceutical Co. This work was supported by the Fonds der Chemischen Industrie.

Table 1: Kinetic Resolution of the Diols *rac*-1a - m

Substrate	R	e.e. of (R)-2 (%)	e.e. of (S)-3 (%)	Conv.	E
1a	Ph-OCH ₂	85	79	0.52	23
1b	2-Me-C ₆ H ₄ -OCH ₂	93	80	0.54	27
1c	3-Me-C ₆ H ₄ -OCH ₂	66	87	0.43	28
1d	4-Me-C ₆ H ₄ -OCH ₂	66	93	0.42	55
1e	2-OMe-C ₆ H ₄ -OCH ₂	63	87	0.42	27
1f	3-OMe-C ₆ H ₄ -OCH ₂	91	95	0.49	>100
1g	4-OMe-C ₆ H ₄ -OCH ₂	96	94	0.51	>100
1h	2-Cl-C ₆ H ₄ -OCH ₂	55	88	0.38	27
1i	3-Cl-C ₆ H ₄ -OCH ₂	86	92	0.48	67
1j	4-Cl-C ₆ H ₄ -OCH ₂	94	92	0.50	85
1k	1-naphthyl-OCH ₂	42	78	0.35	12
1l	Ph	66	93	0.42	55
1m ^a	Et	49 ^b	94 ^c	0.34	53

^a)Pancreatin was used as lipase. ^b)It corresponds to (S)-2. ^c) It corresponds to (R)-3.

References

- Enzymes in Organic Synthesis Part 14. Part 13.: Schick, H.; Schrötter, E.; Szymanowski, M.; Knoll, A. *J. Prakt. Chem./Chemiker-Ztg.*, submitted for publication.
- Parida, S.; Dordick, J. S. *J. Am. Chem. Soc.* **1991**, *113*, 2253 and references cited therein.
- Cesti, P.; Zaks, A.; Klibanov, A. M. *Appl. Biochem. Biotechnol.* **1985**, *11*, 401.
- a) Janssen A. J. M.; Klunder A. J. H.; Zwanenburg, B. *Tetrahedron* **1991**, *47*, 7409, b) Theil, F.; Ballschuh, S.; Kunath, A.; Schick, H. *Tetrahedron: Asymmetry* **1991**, *2*, 1031.
- a) Laumen, K.; Breitgoff, D.; Seemayer, R.; Schneider, M. P. *J. Chem. Soc. Chem. Commun.* **1989**, 148, b) Pederson, R. L.; Liu, K. K.-C.; Rutan, J. F.; Chen, L.; Wong, C.-H. *J. Org. Chem.* **1990**, *55*, 4897, c) Chen, C.-S.; Liu, Y.-C.; Marsella, M. *J. Chem. Soc. Perkin Trans. I* **1990**, 2559, d) Goergens, U.; Schneider, M. P. *J. Chem. Soc. Chem. Commun.* **1991**, 1064, e) Goergens, U.; Schneider, M. P. *J. Chem. Soc. Chem. Commun.* **1991**, 1066, f) Kim, M.-J.; Choi, Y. K. *J. Org. Chem.* **1992**, *57*, 1605.
- Nelson, W. L.; Wennerstrom, J. E.; Sankar, S. R. *J. Org. Chem.* **1977**, *42*, 1006.
- a) Bevinakatti, H. S.; Banerji, A. A. *J. Org. Chem.* **1991**, *56*, 5372, b) Ader, U.; Schneider, M. P. *Tetrahedron: Asymmetry* **1992**, *3*, 521.
- Bianchi, D.; Bosetti, A.; Cesti, P.; Golini, P. *Tetrahedron Lett.* **1992**, *33*, 3231.
- Guo, Z.-W.; Wu, S.-H.; Chen, C.-S.; Girdaukas, G.; Sih, C. J. *J. Am. Chem. Soc.* **1990**, *112*, 4942.
- The E values were determined according to: Chen, C.-S.; Fujimoto, Y.; Girdaukas, G.; Sih, C. J. *J. Am. Chem. Soc.* **1982**, *104*, 7294.
- Kazlauskas, R. J.; Weissfloch, A. N. E.; Rappaport, A. T.; Cuccia, L. A. *J. Org. Chem.* **1991**, *56*, 2656.

(Received in Germany 7 September 1992)